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Nuclear Chemistry, 'Alpha Decay' by Dr. Anju Agrawal, M.Sc.

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Chemistry 175 Nuclearchemistry CHAPTER 21 REVIEW

Nuclear Chemistry SECTION 4 SHORT ANSWER Answer

the following questions in the space provided. 1. Match each of the following statements with the process(es) to which they apply, using one of the choices below: (1) fission only (3) both

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fission and fusion

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Chapter 21 Nuclear Chemistry Review Answers The n:p ratio for Cr-53 is $\frac{29}{24} = 1.21$; for Mn-55, it is $\frac{26}{25} = 1.04$; for Fe-56, it is $\frac{26}{26} = 1.00$. Positron decay occurs when the

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protons and neutrons. nuclide. An atom identified by the
number of protons

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1919 performed the first nuclear transformation. • The transmutations are sometimes represented by listing in order, the target nucleus, the bombarding particle, the ejecting particle and the product nucleus. • The above equation becomes: ${}_{14}^{27}\text{Al} + {}_2^4\text{He} \rightarrow {}_{17}^{31}\text{P} + {}_1^1\text{H}$

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A nuclear fuel. A fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction. The radioactive isotope is contained in tubes called fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

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