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There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. These three states are quite different. The main difference is in their particles.

Chapter 13: States of Matter  
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The relationship among the solid, liquid, and vapor states (or phases) of a substance in a sealed container are best represented in a single

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graph called a phase diagram  
Phase diagram- gives the temperature and pressure at which a substances exists as solid, liquid, or gas (vapor) 70.

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Vocabulary from Chapter 13 "Phases of Matter" for Mr. Testa's Honors Chemistry class. Still a work in progress. Currently only contains some gas laws and properties.

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Matter137 SECTION 13.1 THE  
NATURE OF GASES (pages  
385–389) This section  
introduces the kinetic  
theory and describes how it  
applies to gases. It defines  
gas pressure and explains  
how temperature is related  
to the kinetic energy of the  
particles of a substance.  
Kinetic Theory and a Model  
for Gases (pages 385–386) 1.

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