

Forensics Lab Flame Tests

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[Forensics Lab Flame Tests Where To Download Forensics Lab Flame Tests high temperatures, some electrons may absorb enough energy to allow them to move to higher energy levels. Amy Brown Science: Flame Tests: A Favorite Chemistry Lab Forensic testing is the gathering of data for analysis and for use in legal](#)

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[Flame tests can also be used to determine the presence of metal elements in water by measuring the spectrum produced by the metals exposed to flame. The water is vaporized and then the emissions of the vaporized metals can be analyzed. However, as useful as it is to forensic analysis, the flame test does not work on all elements.](#)

[CSI: Crime Scene Investigation \(Lab Book\): Flame Test](#)

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[Set up the Bunsen burner with a roaring blue flame no more than 5 cm in height. Choose a chemical, record observations about the solution. Predict & record flame color it will produce. Dip one splint in the solution, tap off any excess, and return to your flame station.](#)

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[1. Flame Test. 2. Involves placing the unknown element or compound sample to a flame to determine it's identity through the flame color \(emission spectrum\). The flame would change its color depending on the metal ions present in the given element or compound. The color helps distinguish the given element from the others. Flame test.](#)

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[Light the Bunsen burner with your striker and adjust it so that it has a hot blue flame. Use your tongs to extract a wooden splint from a solution. Make sure you don't dip the tongs into the solution! This will contaminate it! Hold the wooden splint in the hottest part of the burner flame \(the tip of the cone\). Observe the color of the flame. DO NOT catch the splint on fire, because if you do, you can't reuse it to double check the color of the flame test and it will provide a false ...](#)

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[The flame test is used to visually determine the identity of an unknown metal or metalloid ion based on the characteristic color the salt turns the flame of a Bunsen burner. The heat of the flame excites the electrons of the metals ions, causing them to emit visible light.](#)

[How to Do a Flame Test for Qualitative Analysis](#)

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