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Since a sulfur atom

($A_r = 32$) weighs

twice as much as

an oxygen atom

($A_r = 16$), 32 g of

sulfur atoms and

16 g of oxygen

atoms both contain

the same number

of atoms. Answer:

32 g 9.2 (a) 1 mole

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of H atoms = 1 g
1 mole of H₂
molecules = 2 × 1
= 2 g. (b) 1 mole of
N atoms = 14 g

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