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solution is neither acidic or basic.

An acid is a substance that

donates hydrogen ions. Because

of this, when an acid is dissolved

in water, the balance between

hydrogen ions and hydroxide ions

is shifted. Now there are more

hydrogen ions than hydroxide

ions in

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called acidic; if the pH is about 7, the solution is neutral; if the pH is greater than 7, the solution is called basic. In an acidic solution, then, the concentration of hydrogen ions is greater than the concentration of hydroxide ions.

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Indicators are used to determine whether a solution is acidic or alkaline. Acids react with metals, bases and carbonates to produce salts. Neutralisation is the reaction between an acid and a base.

Acidic and alkaline solutions - Acids, alkalis and salts ...

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