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Ideal - Density, Molar Mass,
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Explaining the Gas Laws in
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Answer Key Gas*

An ideal gas is a hypothetical gas consisting of identical particles with zero volume and with no intermolecular forces. Additionally, the constituent atoms or molecules undergo perfectly elastic collisions with the walls of the container. An ideal gas can consist of

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Variables (e.g. carbon dioxide molecules, CO_2) or atoms (e.g. neon atoms, Ne). Real gases do not exhibit these exact prop-

Honors Chemistry POGIL: The ABCs of Gases Unit 06 - Gases ...

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Page 5 . Unit Dimensional
Analysis Activity 10. Here
are 3 other ratio
relationships that we can
obtain from the model: 1
bathroom break . 3 gallons
27 songs 90 miles 75 minutes
\$12.00 . Write 4 other such
relationships that you can
obtain from the model:

Chemistry POGIL Activity «Activity

2 POGIL™ Activities for High

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School Chemistry 1. In Model 1, what does a dot represent? 2. Name two materials that the containers in Model 1 could be made from that would ensure that they were “nonflexible?” 3. In Model 1, the length of the arrows represents the average kinetic energy of the molecules in that sample. Which gas variable (P ...

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12. Does a gas discharge tube filled with boron emit the same wavelengths of light as a tube filled with hydrogen? Use evidence from Model 2 to support your answer. 13 hydrogen and boron support this statement? Circle the appropriate word to complete each statement in Questions 14–17. 14. Electrons and proton–electron) each other.

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Variables Pogil Activities

Answer In this activity, you will explore four variables that quantify gases—pressure (P), volume (V), temperature (T), and moles (n) of gas.

Pogil Gas Variables Model 1

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Access PDF Solution Chemistry Pogil $16.04 \text{ g}/8.3428 \text{ L} = 1.92 \text{ g/L}$ 3. A 1.365-g sample of a pure, unknown gas in a 1.000-L vessel at 22.15 °C has a pressure of 965.4 torr.

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Key Question 1. What is the mass of carbon in 16.0 g of methane? 12.0 g C 2. Show the set-up that would be used to determine the percent by mass of carbon in methane in order to arrive at the answer shown in the Model 3 table. 100 16.0 12.0 × g g 3. What information do you need in order to determine the percent composition by

Instructors Guide: Percent
Composition

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Pogil Buffers Answer Key
Chemistry. One buffer system
in the bloodstream of
animals is the carbonic
acid/bicarbonate buffer. $\text{H}_2\text{CO}_3 + \text{H}_2\text{O} \leftrightarrow \text{HCO}_3^- + \text{H}^+$
+ The carbonic acid in
the bloodstream is formed by
the combination of carbon
dioxide gas and water. This
is a reversible process.

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